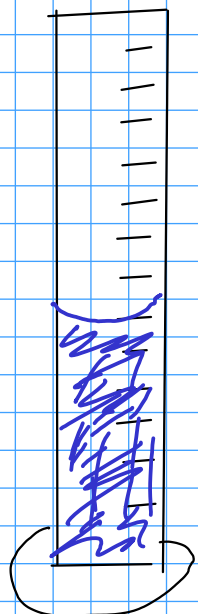


...of an object



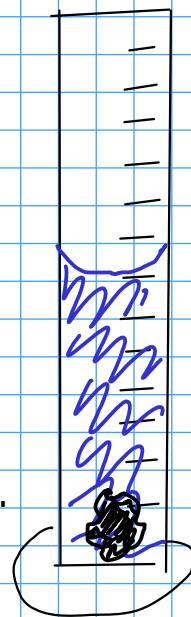
1) Measure mass of object

$$\text{mass} = 9.78 \text{ g}$$



2) Partially fill cylinder with liquid, record volume.

$$\text{volume} = 25.0 \text{ mL}$$



3) Put object into cylinder, record new volume

$$\text{volume} = 26.6 \text{ mL}$$

4) Subtract to find volume of object

$$\begin{array}{r} 26.6 \text{ mL} \\ - 25.0 \text{ mL} \\ \hline 1.6 \text{ mL} \end{array}$$

5) Density = mass object / volume object

$$\text{Density} = \frac{9.78 \text{ g}}{1.6 \text{ mL}}$$

$$= 6.1 \text{ g/mL}$$

Converting from one unit to another

We will use the method of dimensional analysis, sometimes called the factor-label method... or, the "drag and drop" method!

Dimensional analysis uses conversion factors to change between one unit and another

What's a conversion factor? A simple equality.

Example

$$12 \text{ in} = 1 \text{ ft}$$

Conversion factors in metric

In the metric system, conversion factors between units may always be made from the metric prefixes!

For example, "kilo-" means 10^3

$$K = 10^3$$

so

$$K_{\underline{m}} = 10^3 \underline{m}$$

$$K_{\underline{g}} = 10^3 \underline{g}$$

$$K_{\underline{L}} = 10^3 \underline{L}$$

$$K_{\underline{s}} = 10^3 \underline{s}$$

Just apply the prefix to the base unit!

How do we actually USE a conversion factor?

Convert 15.75 m to cm

$$15.75 \cancel{\text{m}} \times \frac{\text{cm}}{10^{-2} \cancel{\text{m}}} = 1575 \text{ cm}$$

$$1 \text{ cm} = 10^{-2} \text{ m}$$

DRAG
AND
DROP!

Put what you want to cancel on
the bottom, then ...

... put what it equals on the top

Convert 0.01893 kg to g

$$0.01893 \cancel{\text{kg}} \times \frac{10^3 \text{ g}}{\cancel{\text{kg}}} = 18.93 \text{ g}$$

$$1 \text{ kg} = 10^3 \text{ g}$$

Convert 14500 mg to kg

$$\text{mg} = 10^{-3} \text{g}$$

$$\text{kg} = 10^3 \text{g}$$

$$14500 \cancel{\text{mg}} \times \frac{10^{-3} \cancel{\text{g}}}{\cancel{\text{mg}}} \times \frac{\text{kg}}{10^3 \cancel{\text{g}}} = 0.0145 \text{ kg}$$

Convert 0.147 mm to μm

$$\text{mm} = 10^{-3} \text{m}$$

$$\mu\text{m} = 10^{-6} \text{m}$$

↖ micro-

$$0.147 \cancel{\text{mm}} \times \frac{10^{-3} \cancel{\text{m}}}{\cancel{\text{mm}}} \times \frac{\mu\text{m}}{10^{-6} \cancel{\text{m}}} = 147 \mu\text{m}$$

Convert 38.47 in to m, assuming 2.54 cm = 1 in

$$2.54 \text{ cm} = 1 \text{ in}$$

$$\text{cm} = 10^{-2} \text{ m}$$

$$38.47 \text{ in} \times \frac{2.54 \text{ cm}}{1 \text{ in}} \times \frac{10^{-2} \text{ m}}{\text{cm}} = 0.9776 \text{ m}$$

Even though English units are involved, we can solve this problem the same way we solved the previous problem where only metric units were used!