## CHM 110 Practice Set 1 Significant Figures and Unit Conversions

Follow the directions for each part, and write answers in the provided blanks. Remember to show work if asked.

Count the number of significant figures in each measurement					
1) 0.00645 g 2)	1050 g	3) 11.73 g	4) 0.0140 g		
5) $4.70 \times 10^6$ g 6)	0.0005 g	7) 0.50057 g	8) 65000 g		
9) 650. g 10) _	72 g				
Solve the following unit conversion problems using dimensional analysis. Write the answer in the answer blank, and show the dimensional analysis setup in the space provided.					
11) 11.3 mL to μL	Answer:		μL		
12) 15500 feet to furlongs. (You ma	ny assume that <i>j</i>	$furlong = 220 \ yd$ and.	3 ft = yd)		
	Answer:		furlongs		
13) 12.4 mg to kg	Answer:		kg		
			2		
14) 0.00045 m <sup>3</sup> to cm <sup>3</sup>	Answer:		cm <sup>3</sup>		

_	te your answer in the blank, and show your work in the provided the significant figures rules we discussed in class.	1 space.
has a mass of 75.2435g.	alculate the density of a liquid, an empty cylinder is weighed. The cylinder filling the cylinder with 25.0 mL of liquid, the cylinder and liquid. What is the density of the liquid?	•
Answer:	<u>g</u>	
	IIIL	
_	te your answer in the blank, and show your work in the provided the significant figures rules we discussed in class.	d space.
,	hass of 7.7548g. It is placed into a graduated cylinder that already cohe object sinks to the bottom of the cylinder, the cylinder reads 17.1 object?	
Answer:	<u>g</u>	
	IIIL	
	sion. Write your answer in the blank, and show your work in th the answer using the significant figures rules we discussed in cla	
17) 72.4 cm to ft (You n	ay assume that $12 in = ft$ and $2.54 cm = in$ . These factors are exact.)	)
Answer:	ft	